

# Tritium: A MicroPower Source for On-Chip Applications

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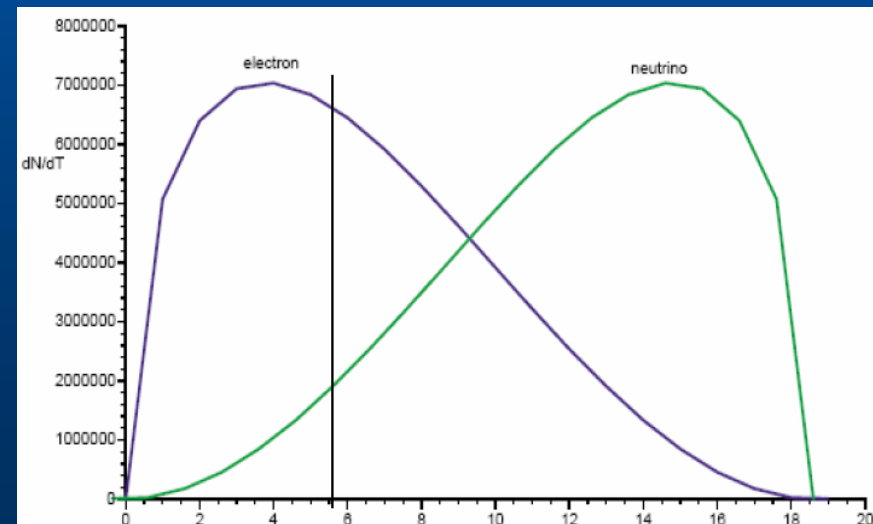
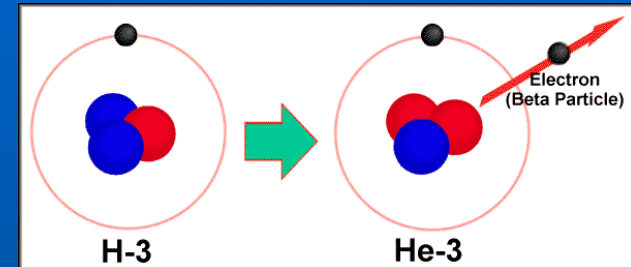
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# Outline

- **Tritium: Basics**
- **Tritium: A MicroPower Source**
  - Beta-Voltaics
  - Beta-Powered MEMS
  - Beta-Luminescence
  - Cold Electron Source
- **Tritium: A Characterization/Diagnostic Tool**
  - Tritium Tracer Studies
  - Tritium Effusion Studies
  - Defect Dynamics
  - Particle Sensor Applications
- **Summary**

# Tritium

- Isotope of Hydrogen
- ${}^3\text{H} \rightarrow {}^3\text{He}^+ + \beta^- + \bar{\nu}_e + 18.6 \text{ keV}$
- Nuclear Half-life:  $t_{1/2} = 12.32 \text{ years}$   
 $\lambda = 1.78 \times 10^{-9} \text{ s}^{-1}$
- Activities:  $1 \text{ Ci} = 3.7 \times 10^{10} \text{ Bq}$   
 $1 \text{ Ci} = 0.39 \text{ std cc}$   
 $1 \text{ Ci} = 33.7 \mu\text{W}$
- Biological: Half-life: 10 days  
ALI\*: 80 mCi  
\*Annual Limit on Intake
- Chemically: Identical to  ${}^1\text{H}$   
Mass effect (~3amu)  
Beta catalysis
- Range (max): 4.5 – 6 mm in air  
5 – 7 micron in water

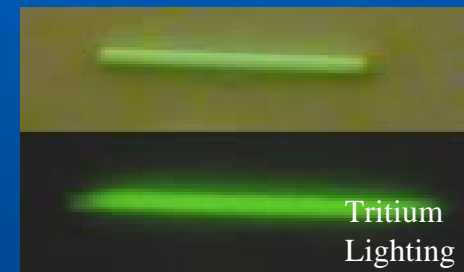
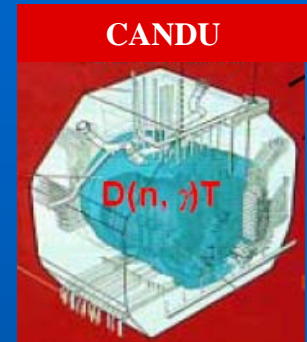


# Producers & Users

## ● Producers of Tritium

- Ontario Power Generation (OPG)
  - ~1 kg/year
- Korean Electric Power Company (KEPCO)
- USA
  - 225 kg produced since 1955
  - 12-75 kg stockpiled
- Russia
- India, Pakistan

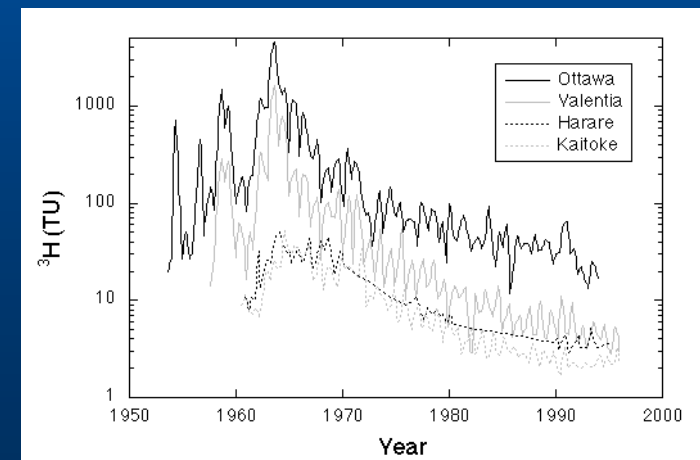
Tritium Producing Burnable Absorber Rods (TPBARs)  
(Lithium Rods in a Light Water Reactor)



## ● Users of Tritium

- Pharmaceutical Research (~100g)
- Tritium Lighting Industry (~30g)
- Fusion Studies
  - Magnetic Confinement (ITER ~40g)
  - Inertial Confinement
- Other

Tritium in  
Natural Waterways



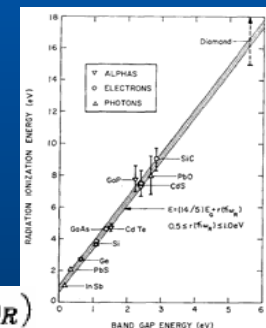
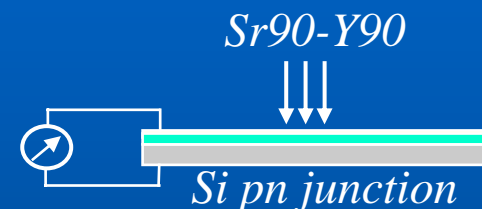
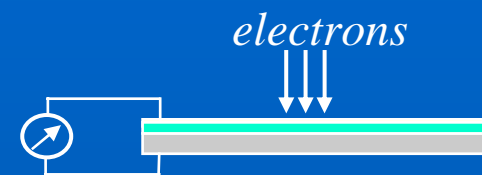
1 Tritium Unit (TU) = 1 T :  $10^{18}$  H

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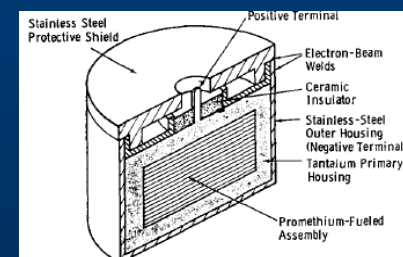
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# Beta Voltaics

- **1951, Ehrenberg, Lang, & West:**
  - Electron-voltaic effect (on a Se device)
- **1956, Rappaport**
  - First direct conversion betavoltaic device (planar configuration, 0.4% efficiency)
- **1968, Klein**
  - Band-gap dependence of electron-hole pair (*ehp*) generation by ionizing radiation
- **1974, Olsen**
  - Theoretical treatment of betavoltaic conversion efficiencies for a variety of semiconductor materials
- **1970s, D W Douglas Laboratories**
  - Planar silicon betavoltaics fueled with  $^{147}\text{Pm}$
  - Efficiencies ranged in 0.7 to 2%



$$\epsilon = (14/5) E_G + r(\hbar\omega_R)$$

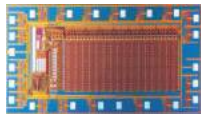


# Renewed Interest in Radioisotope Batteries

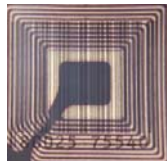
- **Continual miniaturization of electronic and electromechanical systems**
  - Decreased power consumption
- **Integrated Power Sources (SoC)**
- **High energy densities compared to chemical batteries**
- **Operation in extreme environments**
  - For example, temperatures of -100 to +150 °C

# MicroPower Applications

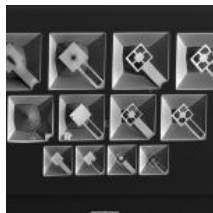
## Sensor/Memory Chips Power requirement: 1-10 $\mu$ W



Non-volatile Memory



RF-ID tag

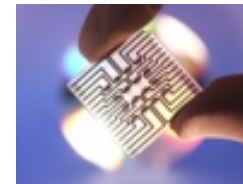


Electrostatic actuation  
of MEMS/NEMS

## SoC Microsystem Power requirement: 1-10 mW



Chip-scale  
atomic clock



Micro-gas  
Analyzer



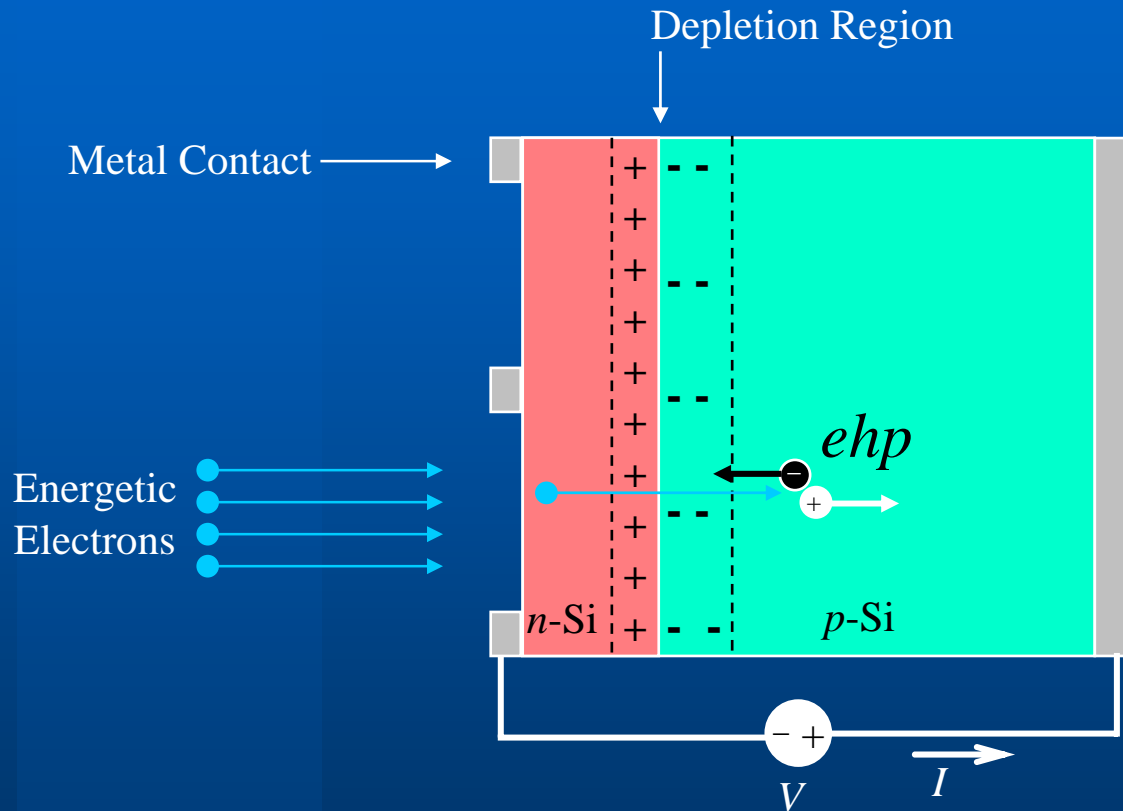
Chip-scale  
Navigation system



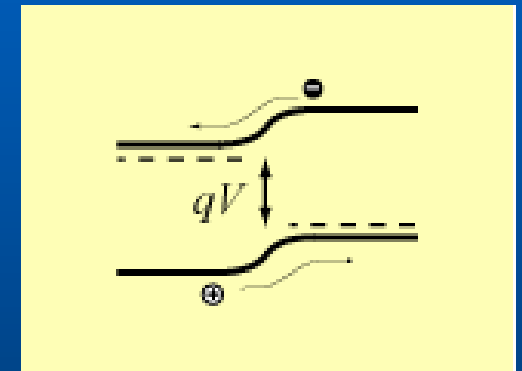
# Market

- **All Batteries:** **\$50 billion**
- **Target markets for betavoltaic batteries**
  - Oil, gas, and environmental
  - Military
  - Medical
  - Space
  - Emerging MEMS/NEMS
- **Market for betavoltaics** **\$1 billion +**

# Electron/Beta Voltaics



Band Diagram



$ehp$ : electron-hole pair

# Choice of Radioisotope

Isotope	E <sub>avg</sub> (keV)	E <sub>max</sub> (keV)	P (W/g)	Work (kWh/ 4y/g)	T <sub>1/2</sub> (yrs)
<b>H-3</b>	5.7	18.6	0.34	10.3	12.3
Ni-63	21	66	0.07	2.5	92
Sr-90	540	900	0.75	25	28
Pm-147	62	230	0.34	7.3	2.6

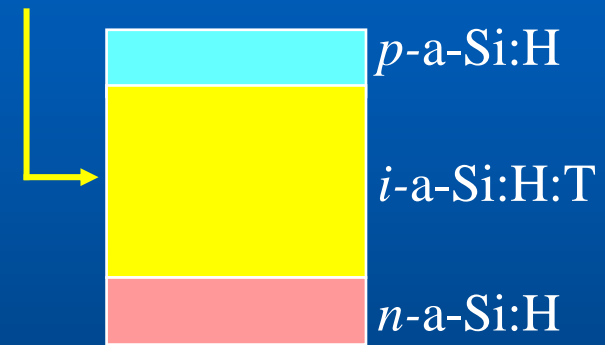
## Tritium

- Low energy  $\beta$ - emitter (benign radioisotope)
- Low cost: \$2.5-\$4/Ci
- Long enough lifetime
- Can be immobilized in a solid matrix
- On-chip integration
- Mature (existing tritium lighting industry)

# Intrinsic Tritiated Amorphous Silicon Betavoltaic Device

- **Substitute tritium for hydrogen in hydrogenated amorphous silicon *pin* photovoltaic devices**
- **Tritium within the energy conversion layer**
  - In contrast to betas originating from a source external to the device
- **Volume source battery**
  - Attained through stacking of many cells
  - In contrast to a planar surface source battery

## Tritiated Intrinsic Layer (uniform)



# a-Si:T Betavoltaic Device

At  $t \sim 0$

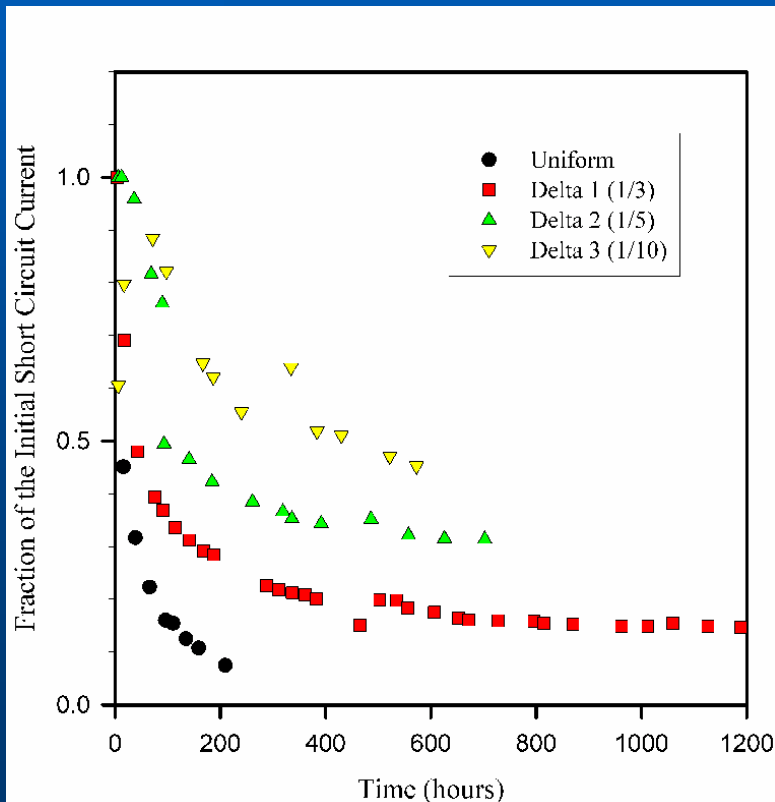
$$I_{sc} = 0.98 \text{ nA}$$

$$V_{oc} = 21 \text{ mV}$$

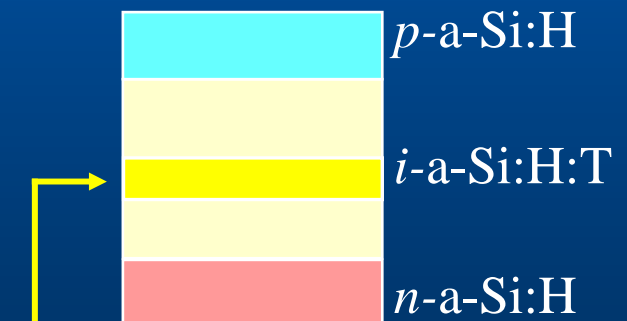
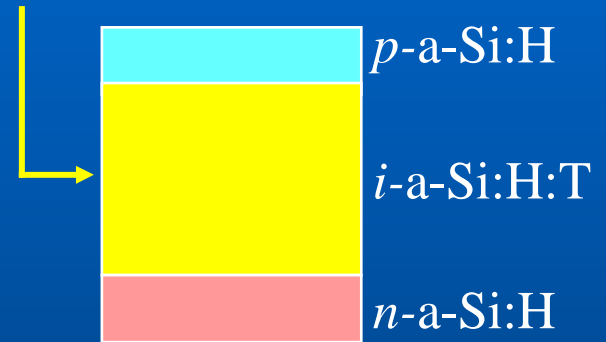
$$\eta = 0.1\%$$

At  $t \sim 10 \text{ days}$

$$I_{sc} < 0.1 \text{ nA}$$



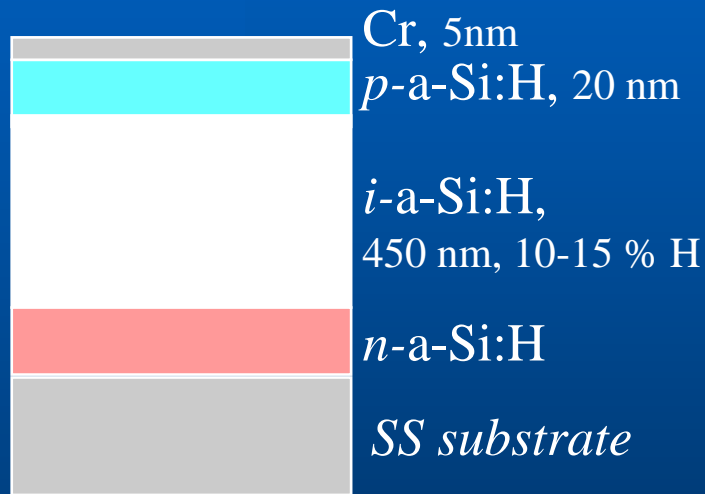
**Tritiated  
Intrinsic Layer (uniform)**



**Tritiated  
Delta Layer**

# a-SiH Betavoltaic Cell Powered by T<sub>2</sub> Gas

## a-SiH Betavoltaics with ultrathin contact



*Tritium gas  
pressure: 678 torr*

At  $t \sim 0$

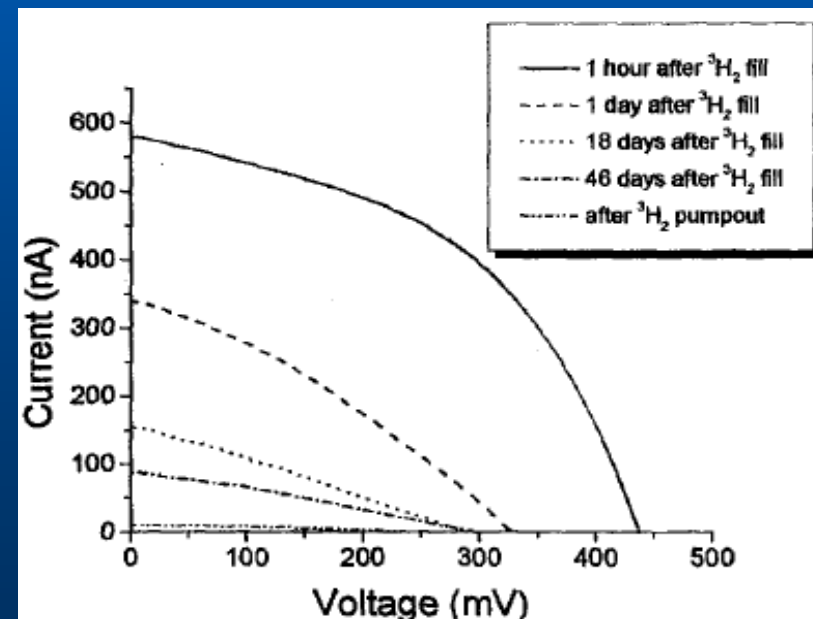
$$I_{sc} = 637 \text{ nA/cm}^2$$

$$V_{oc} = 457 \text{ mV}$$

$$\eta = 1.2\%$$

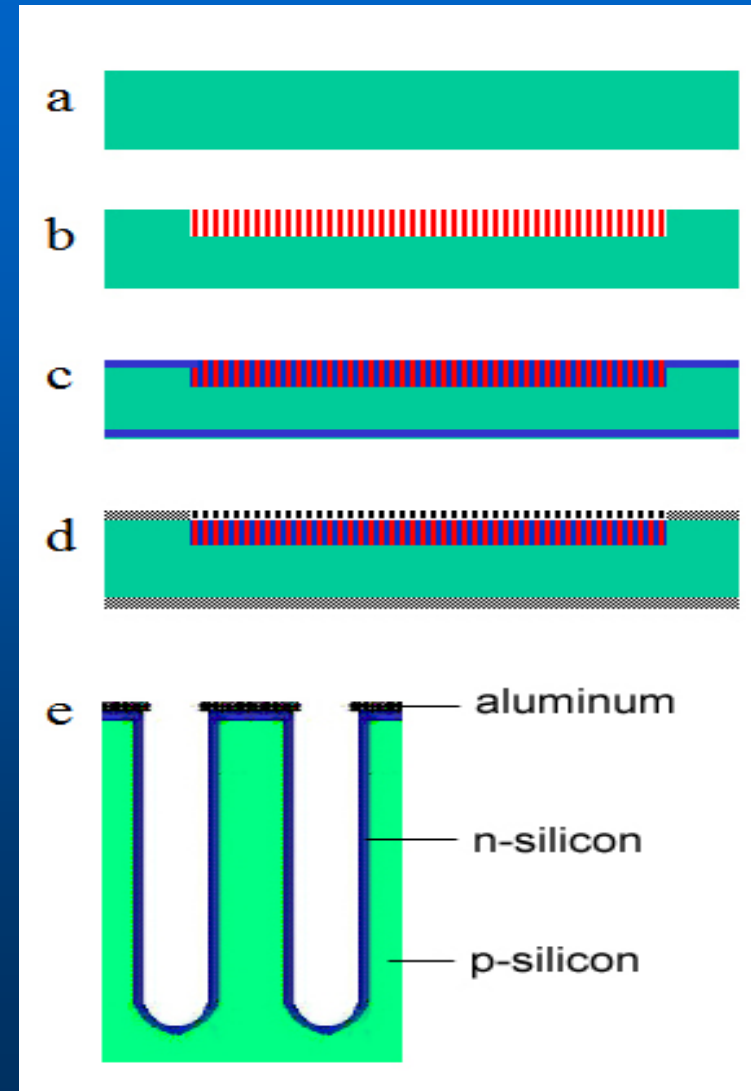
At  $t \sim 46 \text{ days}$

$$\eta < 0.1\%$$

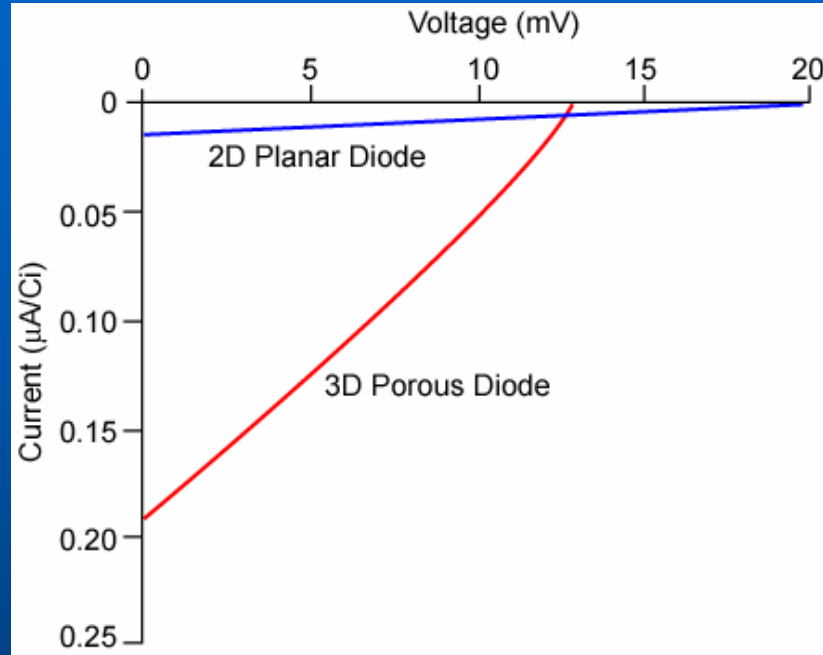
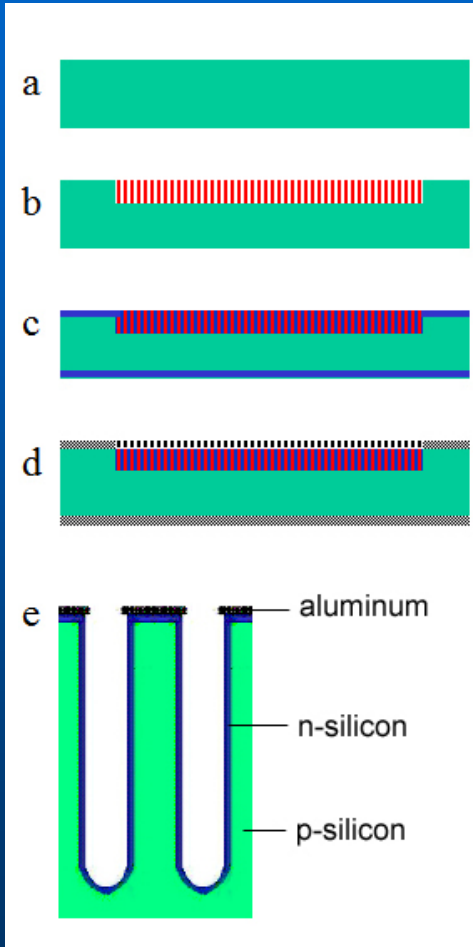


# Porous Silicon 3D Betavoltaics

- Introduce micropores in silicon through electrochemical anodization
- Create  $pn$  junction in the pores through diffusion of n-type dopant
- Introduce an appropriate radionuclide in the pores
- A Volume Source Battery

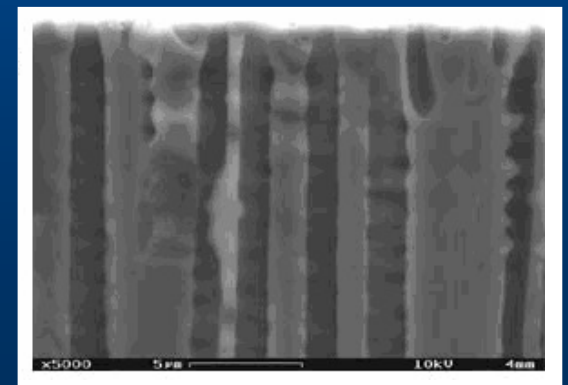
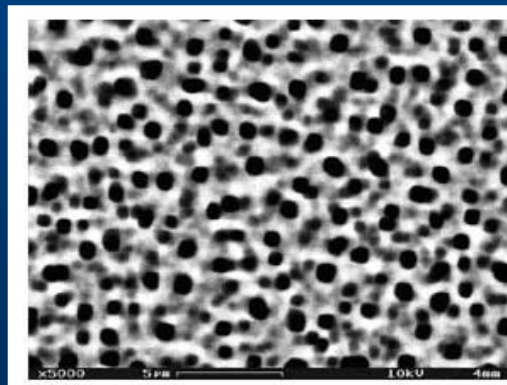


# 3D Versus 2D Betavoltaics



$$\eta_{2D} = 0.02\%$$

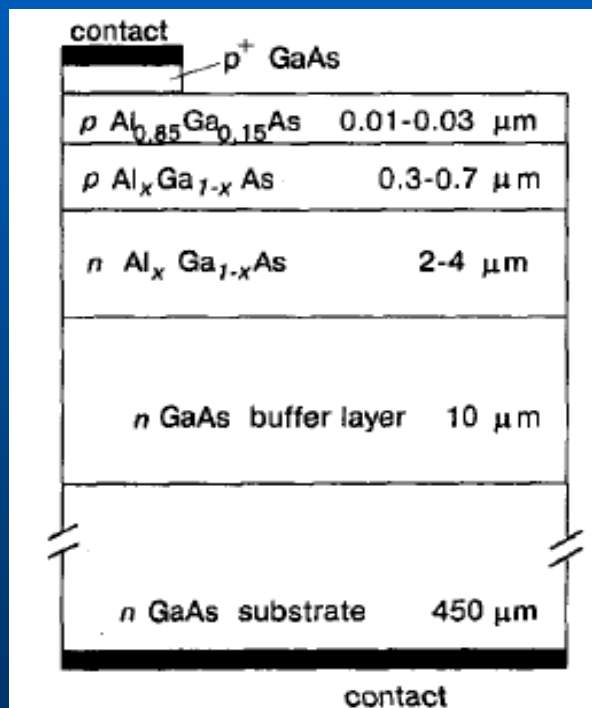
$$\eta_{3D} = 0.2\%$$





# III-V Betavoltaics

## AlGaAs/GaAs Heterojunction Betavoltaics



Source of betas	Generate d current density μA/cm <sup>2</sup>	Open circuit Voltage, V	Output Power, μW/cm <sup>2</sup>	Efficiency (%)
Tritium-titanium	0.04	0.75	0.024	5.6
Tritium gas	0.76	0.91	0.55	5.8
Tritium green lamp	0.12	0.78	0.074	---

# Silicon Carbide Betavoltaics

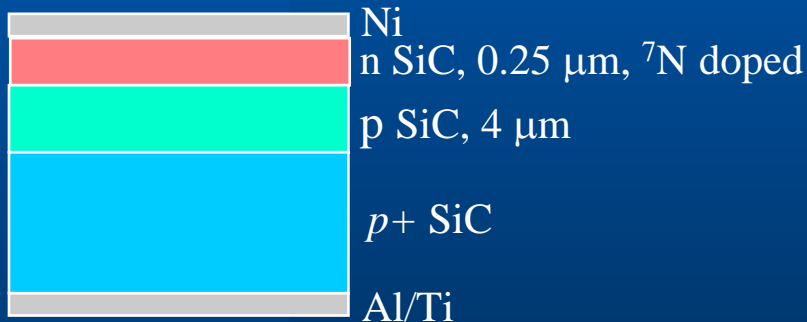
## 4H SiC BV Cell

1 mCi,  $^{63}\text{Ni}$  Source (66keV)

$$I_{\text{sc}} = 16.8 \text{ nA/cm}^2$$

$$V_{\text{oc}} = 0.72 \text{ V}$$

$$\eta = 6\%$$



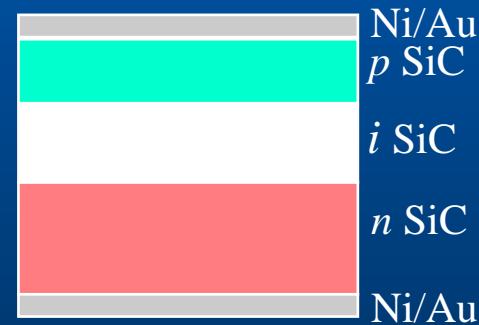
## 4H SiC *pin* BV Cell

8.5 GBq,  $^{33}\text{P}$  Source (249 keV)

$$I_{\text{sc}} = 2.1 \text{ } \mu\text{A/cm}^2$$

$$V_{\text{oc}} = 2.04 \text{ V}$$

$$\eta = 4.5\%$$

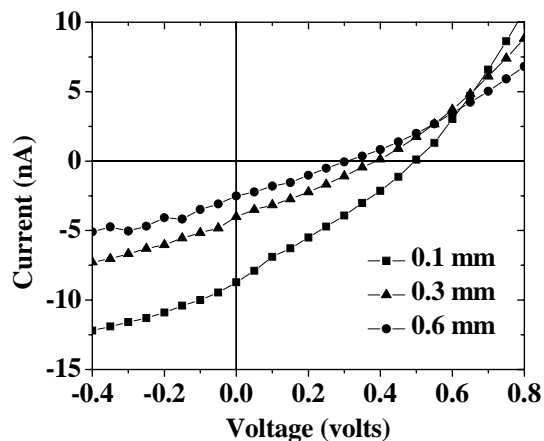
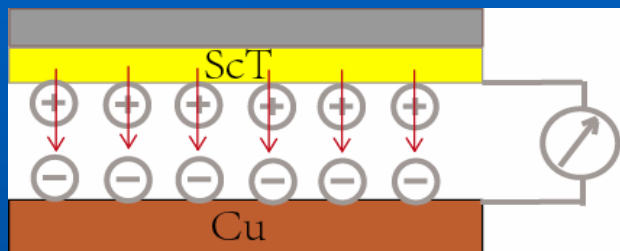


# Contact Potential Difference Betavoltaics

Air-medium CPD BV

$$I_{sc} = 2.7 \text{ nA/cm}^2$$

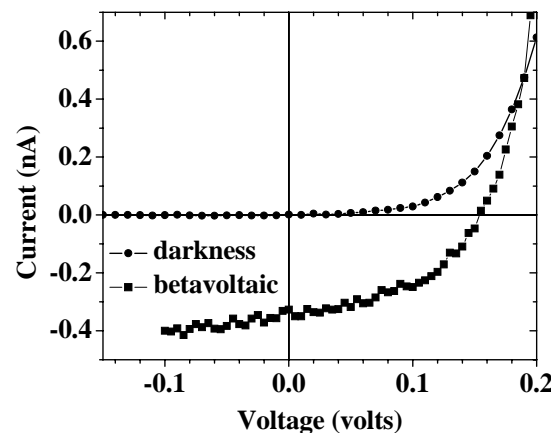
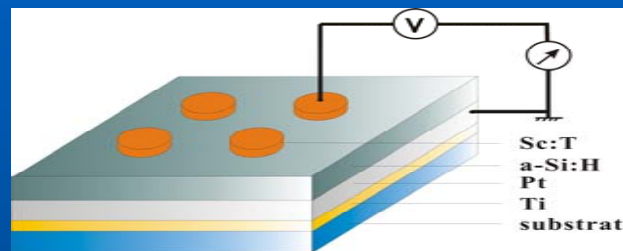
$$V_{oc} = 0.5 \text{ V}$$



Solid CPD BV

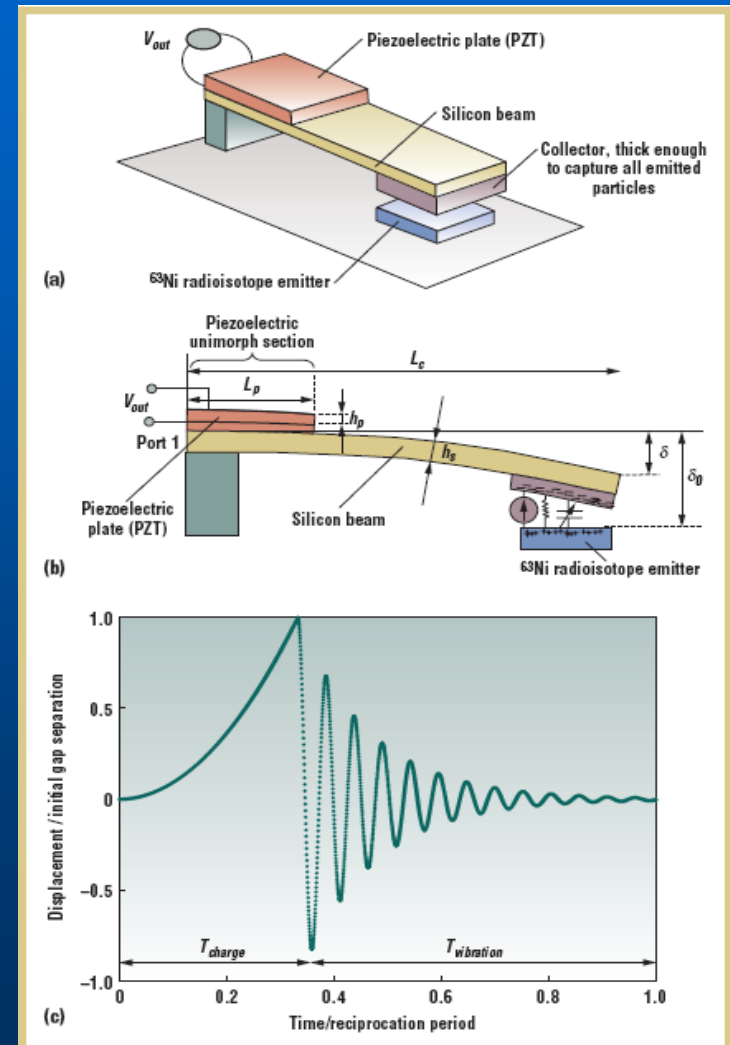
$$I_{sc} = 5.3 \text{ nA/cm}^2$$

$$V_{oc} = 0.16 \text{ V}$$



# MEMS: Radioisotope-Powered Piezoelectric Generator

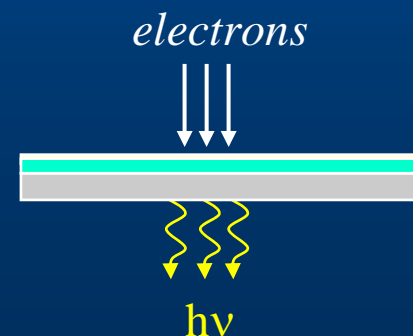
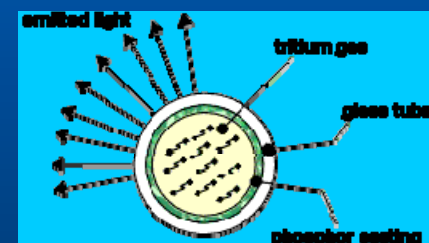
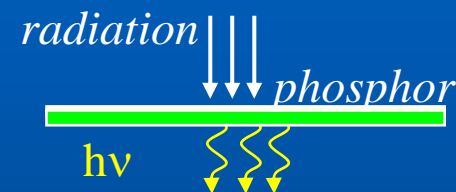
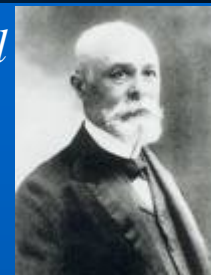
- Self-reciprocating direct-charging cantilever
- Direct conversion of collected-charge-to-motion energy into electrical
  - Radioisotope kinetic energy stored in the cantilever
  - Piezoelectric generator converts stored mechanical energy into electrical energy
- Overall efficiency 2.78%



# BetaLuminescence

- **1898, Becquerel**
  - Radioluminescence
  - Phosphorescence material: potassium uranyl sulphate
- **1920s, Elster, Geitel, and Cookers**
  - Alpha radiation induced scintillations in ZnS.
- **1967, International Atomic Energy Agency (IAEA)**
  - Standards for the use of common RL sources.
  - Most common: tritium beta-luminescence
- **Present**
  - Tritium gas lighting
  - Radium ZnS:Cu paint
  - Novel materials & technologies in Betaluminescence
    - Organic
      - all-organic formulation: polystyrene and fluorescent dye
      - organic system with inorganic phosphor
    - Inorganic
      - semiconductor pn junctions
      - incorporation of tritium in solid matrix: amorphous materials, hydrides, carbon nanotubes, zeolites

Becquerel



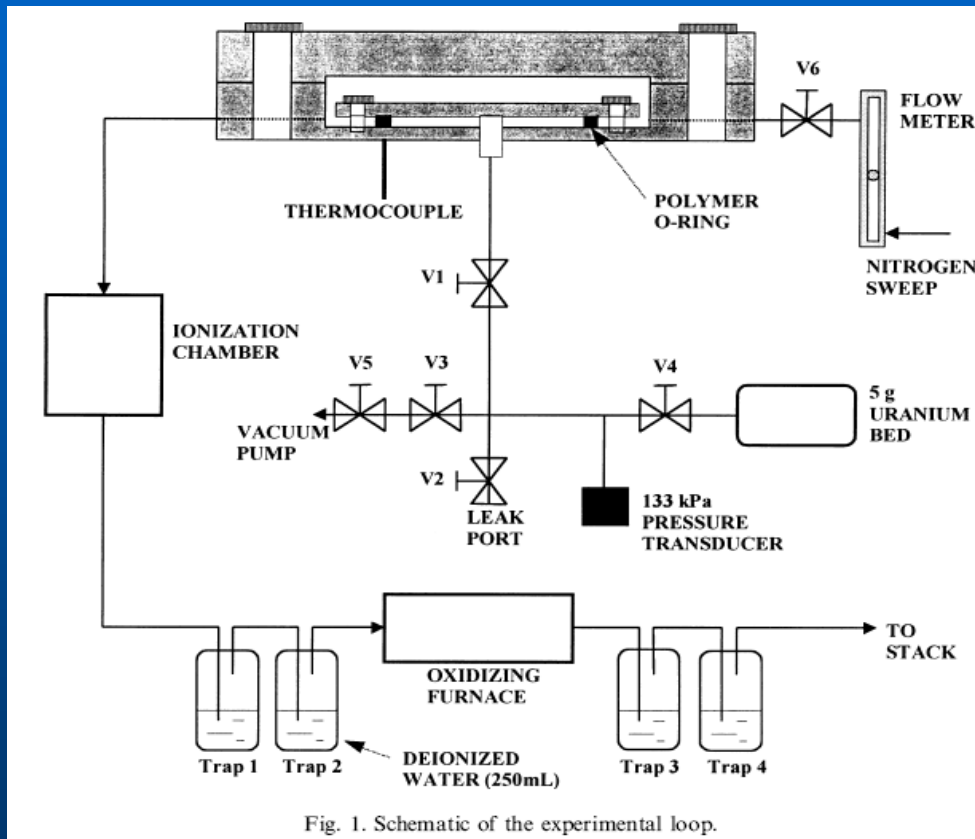
# Cold Electron Source

- **Tritium immobilized in a solid**
- **Materials**
  - **Tritiated metal tritides**
  - **Tritiated amorphous silicon**
    - Plasma enhanced chemical vapour deposition: entire film
    - Tritiation post film deposition: ~50 nm
  - **Tritiated silica on Si-chip**
    - High pressure tritium loading
    - Laser irradiated locked tritium
  - **Tritiated silicon**
    - High pressure tritium loading
    - Surface region: ~ 10 nm
  - **Tritiated carbon nanotubes**

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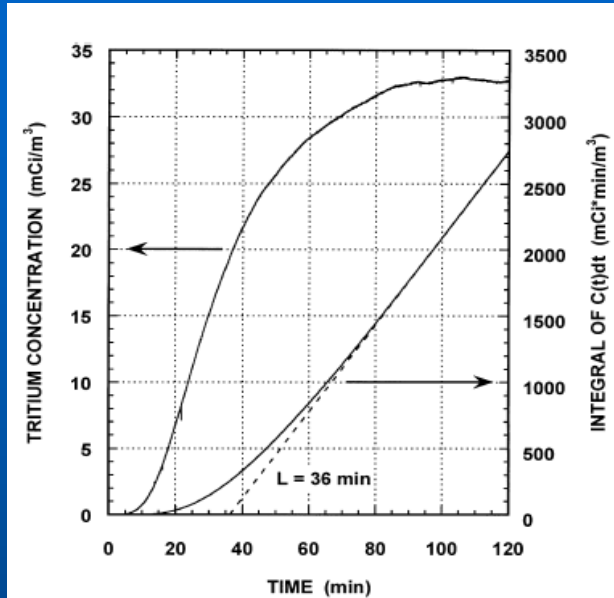
# Tritium Tracer Technique



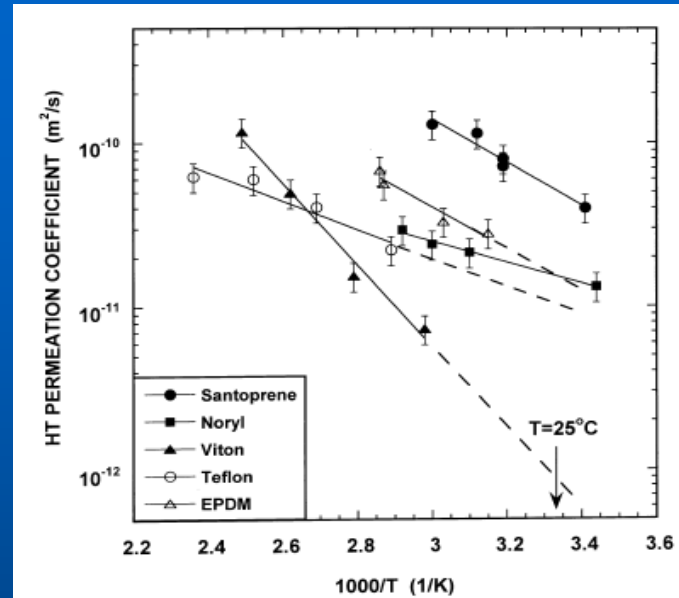
- Tritium as a tracer in measurement of hydrogen permeation in polymer for selection of new material in hydrogen fuel cell.
- Two diagnostics to trace permeating HT: an ionization chamber tritium detector and an HTO water trap/copper oxide furnace/HTO water trap system
- Tritium radiotracer method: simple, effective, reliable.



# Tritium Tracer Technique (cont'd)



Characteristic permeation curve for Noryl at 60 °C



Arrhenius plot of tritium permeation for the five polymers

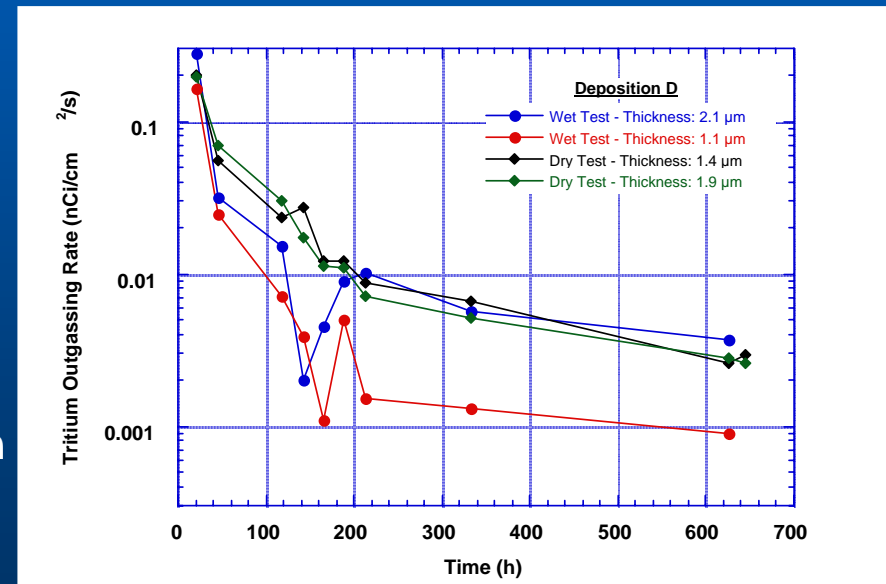
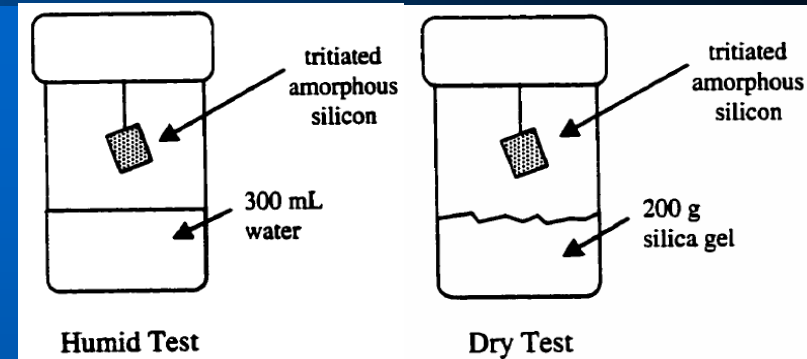
- Materials Tested: EPDM, Teflon, Viton, Santoprene and Noryl
- Permeation Parameters in reasonable agreement with referenced values of H, D, T

Stodilka, Kherani, Shmayda, Thorpe,  
*Intl. J. Hydrogen Energy* 25 (2000) 1129-1136

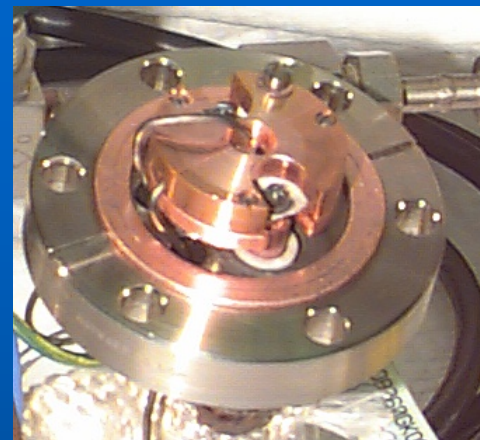
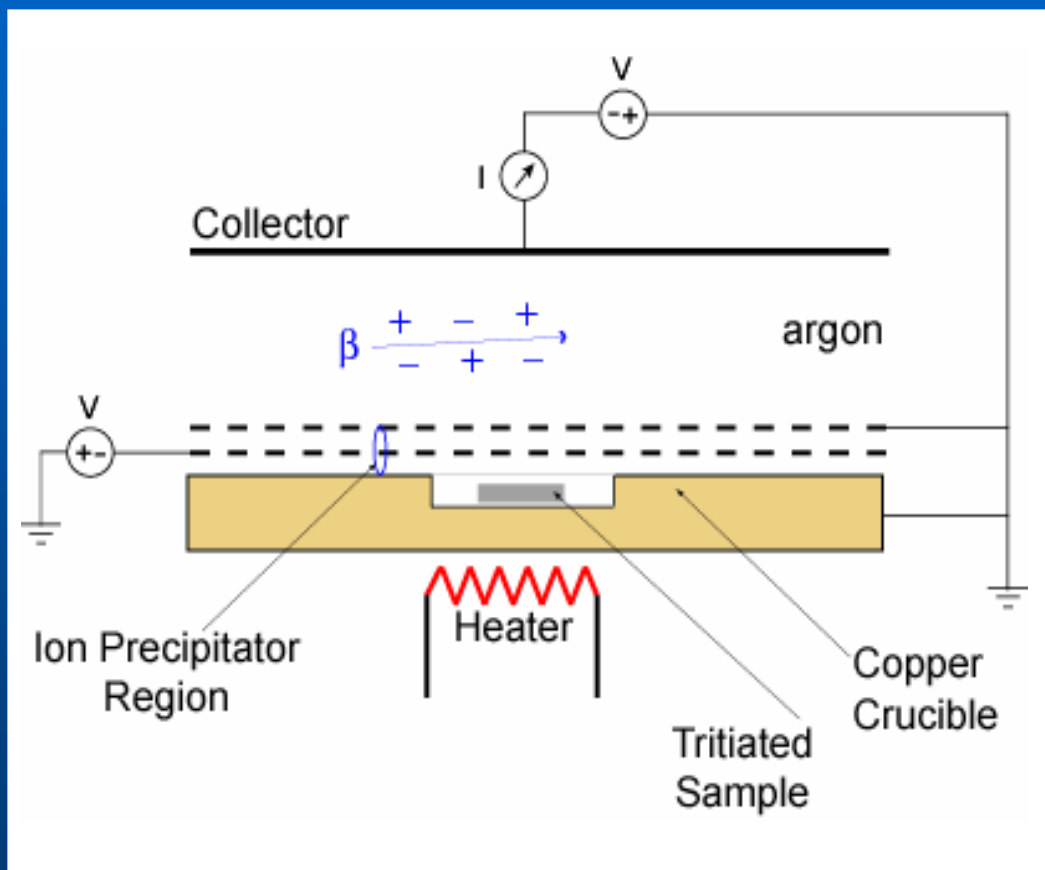
Polymer	Temperature (°C)	$P_0^b$	$E_p^c$	$D_0^b$	$E_d^c$
Viton	63–129	$1.72 \times 10^{-4}$	47.7	$2.22 \times 10^{-5}$	29.1
Teflon	74–150	$8.38 \times 10^{-9}$	16.7	$1.39 \times 10^{-7}$	14.9
EPDM	44–76	$2.74 \times 10^{-7}$	24.4	$3.50 \times 10^{-5}$	27.9
Santoprene	20–60	$1.21 \times 10^{-6}$	25.1	$1.36 \times 10^{-5}$	21.2
Noryl	18–70	$2.11 \times 10^{-9}$	12.3	$4.05 \times 10^{-7}$	16.9

# Tritium Outgassing Studies

- A tool to study hydrogen stability in materials
- High sensitivity
  - Difficult-undetectable for the inactive H-isotope using conventional methods
- Dry and wet test
  - Absorption of HTO desorbed from surface of a given sample
- Tritiated amorphous silicon at room temperature
  - Atomic T concentration: 9%
  - Asymptotic evolution:  $2 \times 10^8 \text{ atm cm}^{-2} \text{ s}^{-1}$
  - Equivalently: Void-Network H diffusion half-life of 60 years
  - This is for a low H stability material, owing to the high void fraction of the material

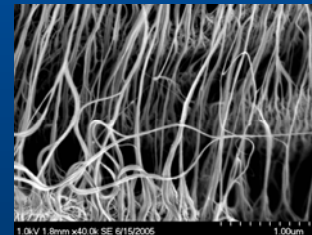
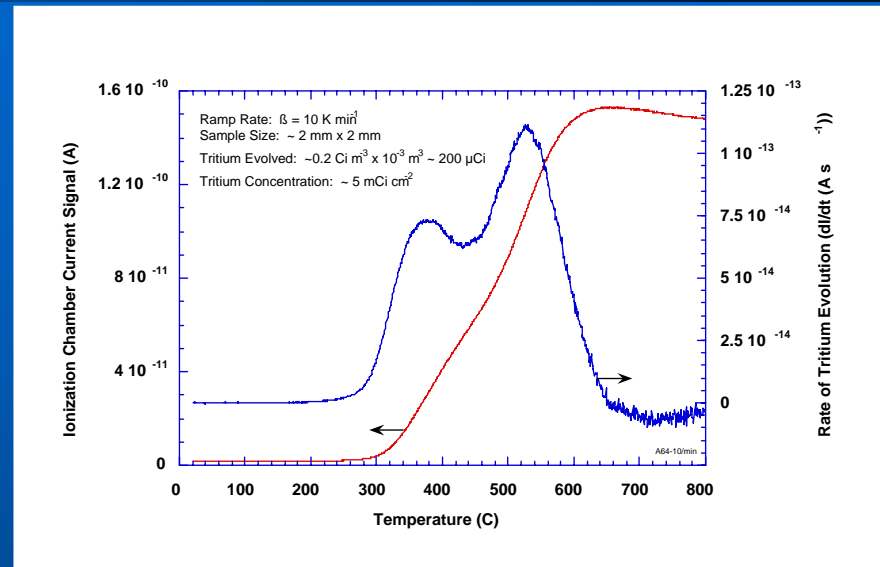


# Tritium Effusion Monitor

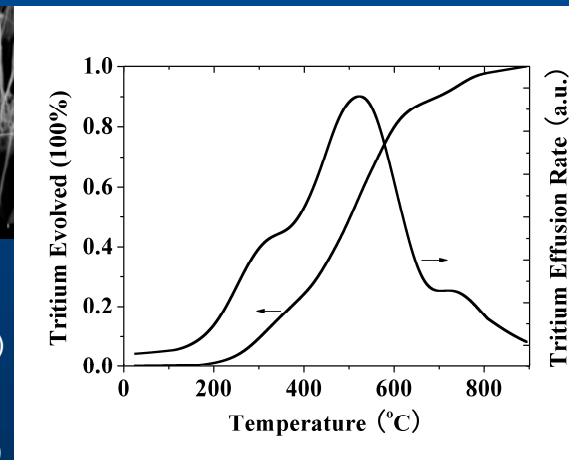


# Tritium Effusion

- **Tritiated amorphous silicon**
  - No tritium evolution at room temperature
  - Characteristic peaks observed at temperatures above the film growth temperature
    - Lower temp peak: higher hydrides SiH<sub>x</sub>
    - Higher temp peak: mono-hydride SiH
- **Tritiated carbon nanotubes**
  - Tritium exposure:
    - 100 bar at 100 °C for 3 days
  - Concentration:
    - Atomic: 1.9%
    - Weight: 0.5%.
  - Gaussian deconvolution:
    - Peaks at 240 °C and 500 °C
    - High temp peak: chemisorbed T
    - Low temp peak: physisorbed T

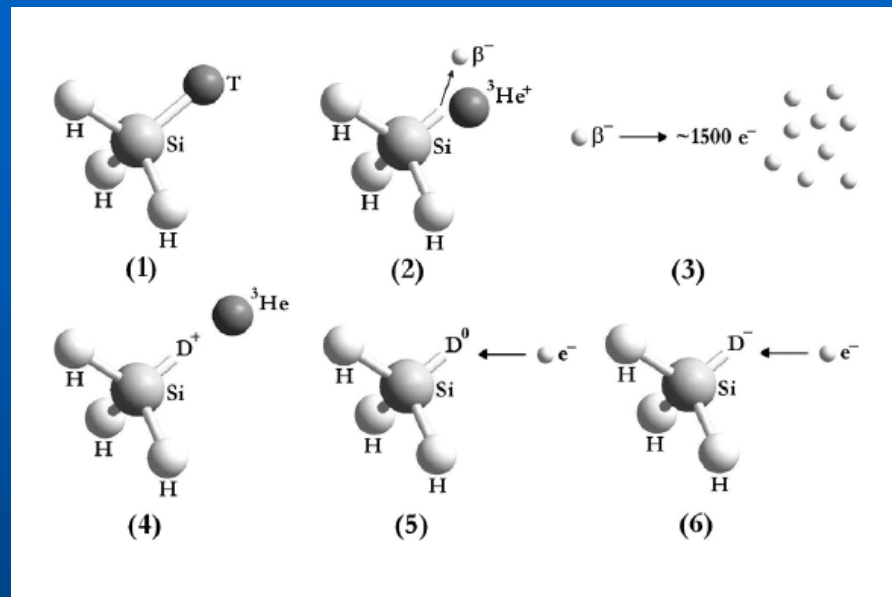


- Purified Single Walled Carbon Nanotubes (SWNT)
- $\sim 25 \mu\text{m}$  paper-like film
- Surface Area:  $1500 \text{ m}^2/\text{g}$
- Density:  $\sim 0.9 \text{ g/cm}^3$ .



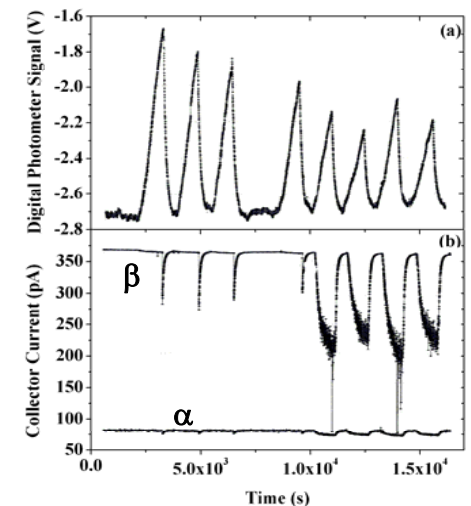
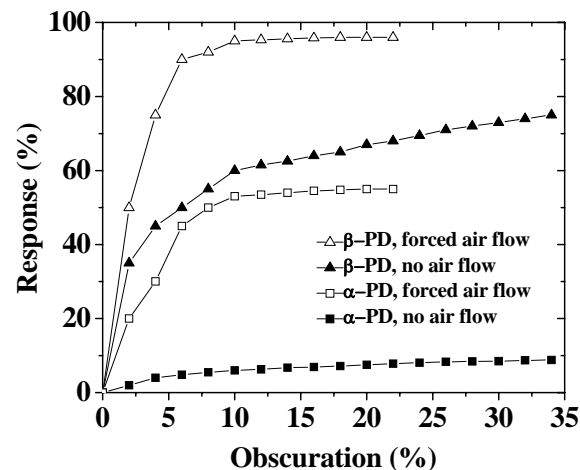
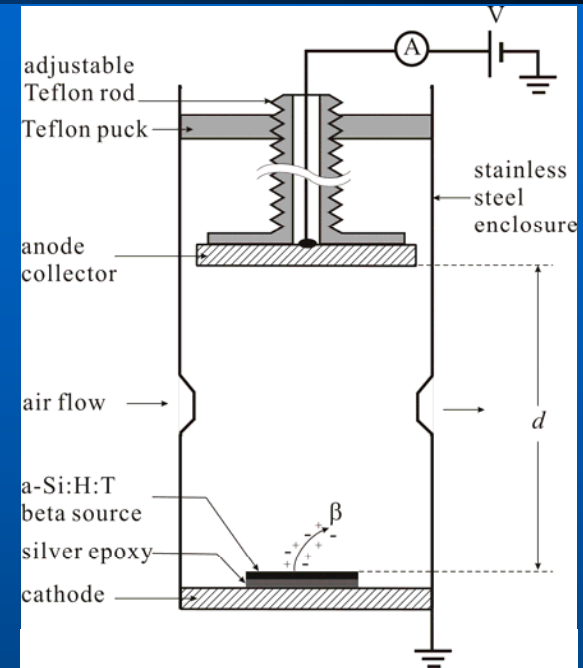
# Defect Dynamics

- **Hydrogenated amorphous silicon solar cells**
  - Staebler-Wronski effect
  - Formation of Si- dangling bonds upon light exposure
  - Drop in efficiency
- **Tritiated amorphous silicon**
  - Defined rate of tritium decay, hence formation of Si-dangling bonds
  - Can study samples under defined conditions (no light exposure)
- **Dynamic defect model**



# Beta Source Particle-Smoke Detector

- Tritium beta source instead of traditional alpha source
  - No gamma emission (as in Am-Be alpha source)
  - Provides bipolar and unipolar regions in the detector
  - Higher absolute current signal
  - Higher sensitivity
    - Several to forty fold more responsive than alpha based detectors
  - Functions like a dual detector (ionization and photoelectric detectors)
    - Smouldering fires
    - Open flame fires



Liu, Alvarez-Ossa, Kherani,  
Zukotynski, Chen,  
*IEEE Sensors J.* 7 (2007) 917.

# Summary

- **Tritium a micro-power source**
  - **Radio-Isotope Micropower Sources (RIMS) is an active area of R&D**
  - **Renewed interest is motivated by continual miniaturization of electronic and electromechanical devices with concurrent reduction in power requirements**
  - **Tritium an amenable radioisotope given its properties and availability**
- **Tritium a powerful diagnostic for hydrogen-material studies**
  - **Ease of experimentation given hydrogen is pervasive**
  - **Unparalleled sensitivity under “non-vacuum” conditions**
  - **Fundamental studies**

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